

Corrigendum

Corrigendum to: “Multistep electron transfer catalytic system for the oxidative carbonylation of phenol to diphenyl carbene” [J. Mol. Catal. A 139 (1999) 109–119]¹

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The run conditions were not reported in the legends of four figures in this article. They may be added as follows:

p. 114. Fig. 1:

Run conditions: $T = 100^{\circ}\text{C}$, $P = 60$ atm ($\text{CO}/\text{O}_2 = 10/1$); $\text{PhOH} = 80$ mmol, $\text{Pd}(\text{OAc})_2 = 0.01$ mmol, $\text{Co}(\text{acac})_3/\text{Pd} = 2/1$ (mol/mol), $p\text{-BQ}/\text{Pd} = 30$ (mol/mol); reaction time = 4–7 h.

p. 115. Fig. 2:

Run conditions: $T = 100^{\circ}\text{C}$, $P = 60$ atm ($\text{CO}/\text{O}_2 = 10/1$); $\text{PhOH} = 80$ mmol, $\text{Pd}(\text{OAc})_2 = 0.01$ mmol, $\text{TBAB}/\text{Pd} = 60/1$ (mol/mol); reaction time = 4 h.

p. 117. Fig. 3:

Run conditions: $T = 100^{\circ}\text{C}$, $P = 60$ atm ($\text{CO}/\text{O}_2 = 10/1$); $\text{PhOH} = 80$ mmol, $\text{Pd}(\text{OAc})_2 = 0.01$ mmol, $\text{TBAB}/\text{Pd} = 60/1$ (mol/mol), $p\text{-BQ}/\text{Pd} = 30$ (mol/mol); $\Delta = 100^{\circ}\text{C}$, reaction time = 5 h; $\circ = 140^{\circ}\text{C}$, reaction time = 1 h.

p. 117. Fig. 4:

Run conditions: $P = 60$ atm ($\text{CO}/\text{O}_2 = 10/1$); $\text{PhOH} = 80$ mmol, $\text{Pd}(\text{OAc})_2 = 0.01$ mmol, $\text{Co}(\text{acac})_3/\text{Pd} = 2/1$ (mol/mol), $\text{TBAB}/\text{Pd} = 60/1$ (mol/mol), $p\text{-BQ}/\text{Pd} = 30$ (mol/mol); reaction time = 1 h.

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¹ PII of original article: S1381-1169(98)00190-3.